

## 2-Acetyl-4-(4-ethoxyphenyldiazenyl)phenol

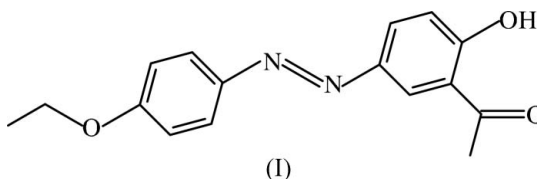
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## Key indicators

Single-crystal X-ray study  
 $T = 296$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å  
 $R$  factor = 0.053  
 $wR$  factor = 0.173  
Data-to-parameter ratio = 15.1For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.In the title structure,  $\text{C}_{16}\text{H}_{16}\text{N}_2\text{O}_3$ , the benzene rings are in a *trans* configuration with respect to the azo double bond and the molecule is essentially planar.Received 12 July 2005  
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## Comment

Azo compounds have been the most widely used class of dyes owing to their versatile application in various fields, such as dyeing textile fibres, colouring different materials, plastics, biological medical studies, lasers, liquid crystalline displays, electrooptical devices and ink-jet printers in high-technology areas (Catino & Farris, 1985; Gregory, 1991). In azo compounds, conversion from the *trans* to the *cis* form can lead to photochromism. Photochromic compounds are of great interest for the control and measurement of radiation intensity, optical computers and display systems (Dürr & Bouas-Laurent, 1990), and for potential applications in molecular electronic devices (Martin *et al.*, 1995). As part of a general study of the crystal chemistry of dyes, and to provide templates for molecular-modelling studies, the crystal structure of the title compound, (I), has been determined.The molecular structure of (I) is shown in Fig. 1 and selected geometric parameters are given in Table 1. The bond lengths and angles of the azo group are as expected. The molecule is essentially planar, with dihedral angles between the mean planes of the benzene rings and the  $\text{C1}-\text{N1}=\text{N2}-\text{C7}$  azo bridge of 6.19 (6) and 4.12 (4) Å for rings C1–C6 and C7–C12, respectively. The angle between the planes of the two benzene rings is 10.14 (4) Å. Apart from the expected intramolecular  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bond (Table 2) there are no other significant hydrogen-bond interactions, the closest contacts being of the types  $\text{C}-\text{H}\cdots\text{O}$  and  $\text{C}-\text{H}\cdots\pi$ , with distances *ca* 2.7 Å for  $\text{H}\cdots\text{O}$  and  $\text{H}\cdots\pi$ (ring centroid).

## Experimental

The compound was prepared from 4-ethoxyaniline and 2-hydroxyacetophenone (Deveci *et al.*, 2005). Crystals of 2-acetyl-4-(4-ethoxyphenyldiazenyl)phenol were obtained after 1 d by slow evaporation of an acetonitrile solution (yield 56%, m.p. 408–410 K).

## Crystal data

C<sub>16</sub>H<sub>16</sub>N<sub>2</sub>O<sub>3</sub>  
*M<sub>r</sub>* = 284.31  
 Orthorhombic, *Pbca*  
*a* = 7.2099 (5) Å  
*b* = 13.6448 (9) Å  
*c* = 29.631 (3) Å  
*V* = 2915.0 (4) Å<sup>3</sup>  
*Z* = 8  
*D<sub>x</sub>* = 1.296 Mg m<sup>-3</sup>

Mo *K*α radiation  
 Cell parameters from 13929 reflections  
 $\theta = 2.8\text{--}27.9^\circ$   
 $\mu = 0.09\text{ mm}^{-1}$   
*T* = 296 K  
 Plate, brown  
 0.80 × 0.39 × 0.06 mm

## Data collection

Stoe IPDS-2 diffractometer  
 $\omega$  scans  
 Absorption correction: integration  
 (*X-RED32*; Stoe & Cie, 2002)  
*T<sub>min</sub>* = 0.951, *T<sub>max</sub>* = 0.994  
 19031 measured reflections  
 2864 independent reflections

1211 reflections with *I* > 2σ(*I*)  
*R<sub>int</sub>* = 0.111  
 $\theta_{\text{max}} = 26.0^\circ$   
*h* = −8 → 8  
*k* = −16 → 16  
*l* = −36 → 36

## Refinement

Refinement on *F*<sup>2</sup>  
*R*[*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.053  
*wR*(*F*<sup>2</sup>) = 0.173  
*S* = 0.89  
 2864 reflections  
 190 parameters

H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0811P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.12\text{ e \AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.20\text{ e \AA}^{-3}$

Table 1

Selected geometric parameters (Å, °).

C1—C6	1.382 (4)	C8—C9	1.373 (4)
C3—C4	1.403 (4)	C10—O1	1.350 (4)
C4—O3	1.359 (3)	C10—C11	1.397 (4)
C7—C12	1.377 (4)	C13—O2	1.242 (4)
C7—C8	1.398 (4)		
C9—C10—C11	121.1 (3)	O3—C15—C16	107.1 (3)
C12—C11—C13	121.9 (3)	C4—O3—C15	118.4 (2)
C11—C13—C14	121.7 (3)		
C1—N1—N2—C7	178.4 (3)	C3—C4—O3—C15	−173.1 (3)

Table 2

Hydrogen-bond geometry (Å, °).

Cg1 is the centroid of the C1—C6 ring.

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
O1—H1...O2	0.82	1.83	2.555 (4)	146
C14—H14C...O1 <sup>i</sup>	0.96	2.70	3.464 (4)	137
C16—H16C...Cg1 <sup>ii</sup>	0.96	2.75	3.617 (4)	149

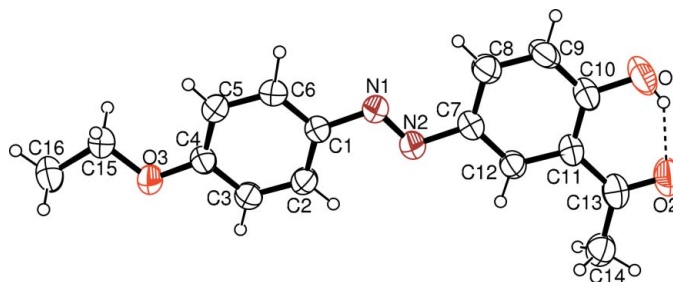
Symmetry code: (i)  $-x + \frac{1}{2}, y - \frac{1}{2}, z$ ; (ii)  $x - \frac{1}{2}, -y + \frac{1}{2}, -z$ .

Figure 1

A view of (I), showing the atom-numbering scheme and 50% probability displacement ellipsoids. The dashed line indicates an intramolecular hydrogen bond

All H atoms were positioned geometrically and refined using a riding model, with C—H = 0.93–0.97 Å and O—H = 0.82 Å. *U<sub>iso</sub>*(H) values were set at 1.2*U<sub>eq</sub>*(C) or 1.5*U<sub>eq</sub>*(O or C<sub>methyl</sub>).

Data collection: *X-AREA* (Stoe & Cie, 2002); cell refinement: *X-AREA*; data reduction: *X-RED32* (Stoe & Cie, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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